

Novel acyclic sulfide, garlicnin L-5, from garlic

Shang-Hui Gao, Toshihiro Nohara*, Tsuyoshi Ikeda, Jian-Rong Zhou and Kazumi Yokomizo

Faculty of Pharmaceutical Sciences, Sojo University, 4-22-1, Ikeda, Nishi-ku, Kumamoto 860-0082, Japan.

ABSTRACT

As part of our continuing studies on the *Allium* sulfides exhibiting antitumor effects, we isolated one new sulfide and characterized its structure as a novel dimer compound combined with the partial structures of allyl, popenyl and 2-formylbutane moieties through the sulfinyl and thiosulphenyl functions.

KEYWORDS: *Allium sativum*, acyclic sulfide, sulfinyl function, thiosulphenyl function, dimer, 2-formyl butane moiety.

INTRODUCTION

Unexpectedly, we found a few clarified sulfides from garlic; very few cyclic sulfides from garlic (*Allium sativum*) [1-10], onion (*A. cepa*) [6, 11-15], and Welsh onion (*A. fistulosum*) had been found before our studies on them [12]. Therefore, we started research aiming at the isolation, structural characterization and pharmacological analysis of the cyclic sulfides from garlic that show antitumor activity. So far, we obtained four acyclic-type sulfides (garlicnins L-1, L-2, L-3, and L-4 [4]), nine 3,4-dimethylthiolane-type sulfides (garlicnins A [1], B₁, B₂, B₃, B₄, C₁, C₂, C₃ [2, 3], and M [7]), four 2-methylthiolane (and thiane)-type sulfides (garlicnins I₁ [5], I₂ [7], J₁ [5], and J₂ [8]), two 1,2-dithiolane-type sulfides (garlicnins G [5], and P [8]), and two 2-oxothiolane-type sulfides (onionins B₁, and B₂ [13]), together with the known sulfides, (*E*)-ajoene [16], and kujounin A₁ derivative [17] from the

acetone extract of garlic. Furthermore, we recently isolated a new bis-thiolane-type compound, garlicnin IB [18]. The structures of these compounds are summarized in Table 1 in which the corrected structures of garlicnins A and B are represented according to the references [19, 20].

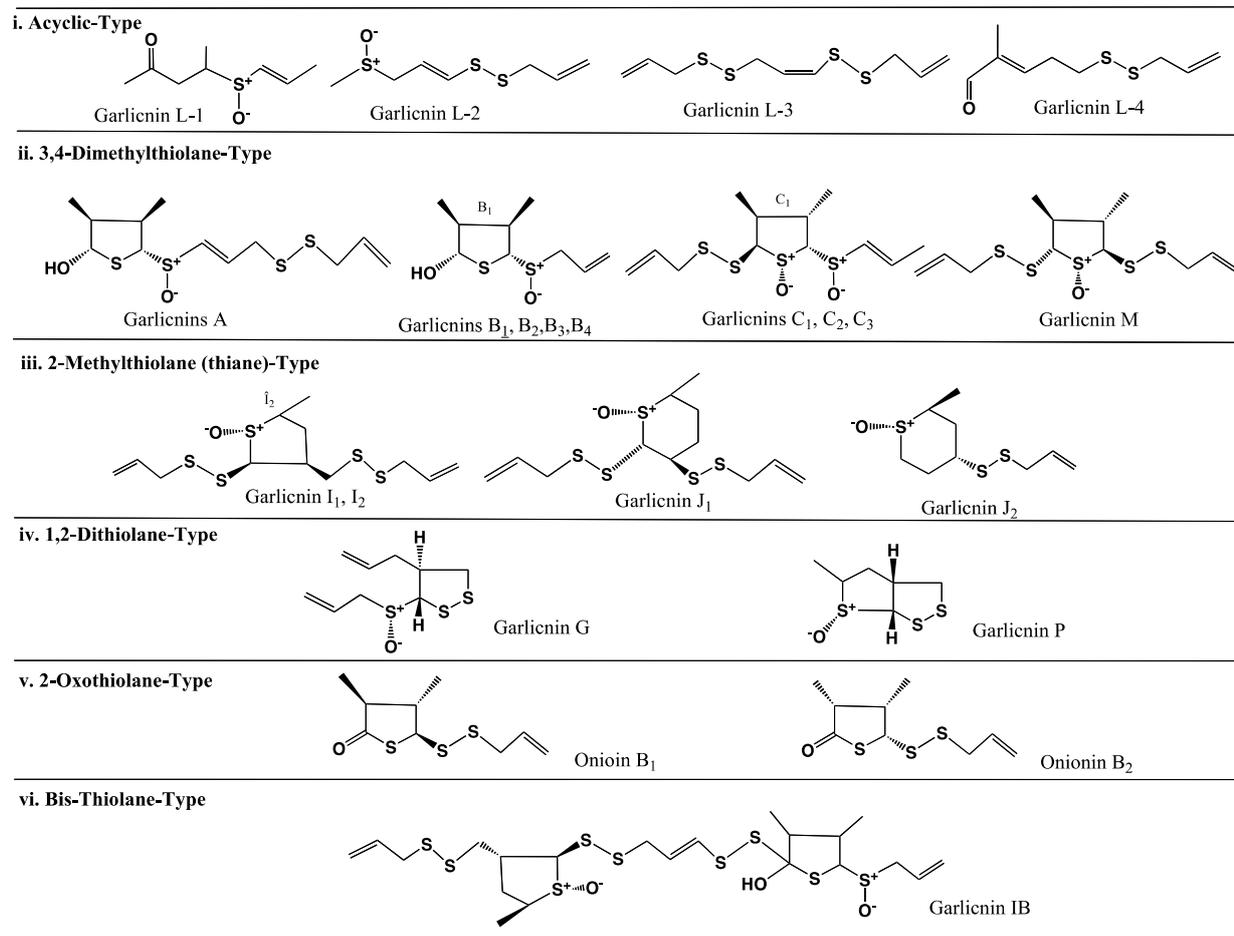
Onionin A₁ obtained from onion, which is the isomer of a major component garlicnin B₁ from garlic, showed potential in inhibiting the polarization of M2-activated macrophages that were capable of suppressing tumor-cell proliferation [21, 22]. The effects of onionin A₁ on tumor progression and metastasis in both osteosarcoma and ovarian cancer-bearing mouse models were examined. Tumor proliferation was depressed, and tumor metastasis was controlled by regulating macrophage activation. These results showed that the 3,4-dimethylthiolane-type sulfide, onionin A₁ was an effective agent for controlling tumors in both *in vitro* and *in vivo* models, and that the antitumor effects observed *in vivo* are likely caused by reversing the antitumor immune system. Activation of the antitumor immune system by onionin A₁ might be an effective adjuvant therapy for patients with osteosarcoma, ovarian cancer and other malignant tumors [14, 15]. Considering this fact, we continue further the experiment to obtain new sulfides. In this study, we deal with the structural characterization of new sulfide **1** from the acetone extract of garlic.

MATERIALS AND METHODS

General experimental procedures

The ¹H- and ¹³C-NMR spectra were measured in CDCl₃ using a JEOL alpha 500 spectrometer at

*Corresponding author: none@ph.sojo-u.ac.jp

Table 1. Garlicinins and onioninins isolated from garlic.

500 and 125 MHz, respectively, and the chemical shifts were found to be on the δ (ppm) scale. Column chromatography was carried out on silica gel 60 (230-400 mesh, Merck). Thin layer chromatography (TLC) was performed on silica gel plates (Kieselgel 60 F254; Merck). TLC spots were visualized under UV light (254/366 nm), sprayed with 10% H₂SO₄, and then heated.

Plant material

We used the Chinese garlic bulbs (*A. sativum* L., family Liliaceae) imported by Shinko Co., Ltd. A voucher specimen (GAR-20-10-36) was deposited in the Herbarium of the Botanical Garden at Sojo University, Kumamoto, Japan.

Extraction and isolation

Peeled Chinese garlic bulbs (Shinko Company, 2268 g) purchased at Kumamoto city were

roughly chopped and homogenized in a mixer along with acetone (4.5 l). The mixture was subsequently soaked in acetone for 3 days at room temperature. The filtrate was concentrated at 40 °C *in vacuo* to give a suspension, which was extracted using ethyl acetate and water. The organic layer was taken and evaporated in reduced pressure at 40 °C to afford the residue (14.85 g), which was then column chromatographed on silica gel (width: 4.8 × length: 35 cm with *n*-hexane:acetone = 5:1 → 4:1 → 3:1 → 2:1) to provide fractions 1 (4.4673 g), 2 (1.1488 g), 3 (0.2607 g), 4 (1.2804 g), 5 (0.2470 g), 6 (0.5459 g), and 7 (1.6801 g). Fraction 2 was mostly composed of garlicin B₁ and fraction 4 was mostly composed of *E*-ajoene [16]. The fraction 6 was then column chromatographed on silica gel with *n*-hexane:acetone = 2:1 to give three fractions:

fractions 0, 1 (78.4 mg), and 2 (279.3 mg). The fraction 2 was further silica gel chromatographed with $\text{CHCl}_3:\text{MeOH} = 150:1 \rightarrow 100:1 \rightarrow 50:1 \rightarrow 25:1$ to afford one compound, named garlicnin L-5 (**1**, 34.8 mg).

Garlicnin L-5 (**1**)

Colorless resinous syrup, *R_f* value: 0.36 ($\text{CHCl}_3 : \text{MeOH} = 20 : 1$); 0.30 (*n*-hexane:acetone = 2:1) on TLC; ^1H - and ^{13}C -NMR spectra; allyl sulfinyl group (**1-a**): olefinic methylene at δ_{H} 5.48 (1H, d, $J = 15.2$ Hz, H-1a), and 5.53 (1H, d, $J = 10.3$ Hz, H-1b); δ_{C} 125.1 (C-1), olefinic methine at δ_{H} 5.45 (1H, m, H-2); δ_{C} 124.7 (C-2), methylene at δ_{H} 3.76 (2H, d, $J = 7.5$ Hz, H₂-3); δ_{C} 58.6 (C-3), 2-formyl-butane thiosulfinyl group (**1-b**): methylene at δ_{H} 3.30 (2H, d, $J = 7.5$ Hz, H₂-5); δ_{C} 47.4 (C-5), methine at δ_{H} 3.07 (1H, t, $J = 7.5$ Hz, H-6); δ_{C} 35.6 (C-6), formyl function at δ_{H} 9.80 (1H, d, $J = 10.3$ Hz, 6-CH=O); δ_{C} 197.2 (6-CH=O), methine at δ_{H} 2.99 (1H, t, $J = 7.5$ Hz, H-7); δ_{C} 36.4 (C-7), methylene at δ_{H} 3.30 (2H, d, $J = 7.5$ Hz, H₂-8); δ_{C} 44.1 (C-8), propenyl group (**1-c**): methyl function at δ_{H} 1.98 (3H, d, $J = 1.75$ Hz, H₃-13); δ_{C} 17.5 (C-13), olefinic methine at δ_{H} 6.96 (1H, q, $J = 6.85$ Hz, H-12); δ_{C} 145.9 (C-12), olefinic methine at δ_{H} 6.32 (1H, d, $J = 16.6$ Hz, H-11); δ_{C} 129.2 (C-11), as shown in Figure 1.

RESULTS AND DISCUSSION

Garlicnin L-5 (**1**) was obtained as a resinous syrup. The $[\text{M}+\text{H}]^+$ peak on the positive high-resolution fast-atom bombardment mass spectroscopy (HR-FAB-MS) was not obtained, but it showed unity on the TLC [*R_f* value: 0.36 ($\text{CHCl}_3:\text{MeOH} = 20:1$); 0.30 (*n*-hexane:acetone = 2:1)]. The spectral analysis by the ^1H - ^1H correlation spectroscopy (COSY), ^1H -detected heteronuclear correlation through multiple quantum coherence (HMQC), and heteronuclear multiple bond correlation (HMBC) revealed the presence of the 3 partial structures; however, these did not connect the 3 substructures. The sulfinyl function or the thiosulfinyl function would make the three substructures separated partial structures. Here, we assigned the NMR chemical shifts by taking the following points into consideration. The methylene carbon adjacent to the sulfinyl function appeared at about δ 47-60; on the other hand, the methylene carbon adjacent to the thiosulfinyl function appeared at around δ 40-45 [2, 3, 7, 18].

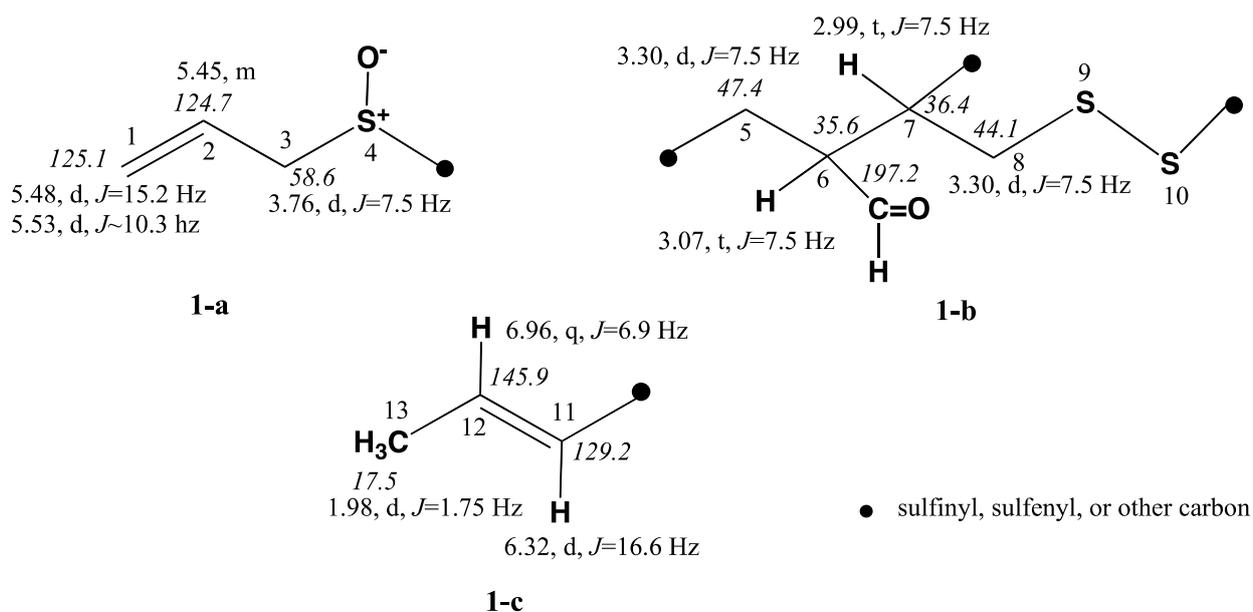


Figure 1. Partial structures of garlicnin L-5 (**1**) with NMR data.

The assignments of NMR data of three portions were made as follows: the first (**1-a**) was allyl sulfinyl group: from olefinic methylene at δ_{H} 5.48 (1H, d, $J=15.2$ Hz), and 5.53 (1H, d, $J=10.3$ Hz); δ_{C} 125.1, next to the olefinic methine at δ_{H} 5.45 (1H, m); δ_{C} 124.7, further to the methylene at δ_{H} 3.76 (2H, d, $J=7.5$ Hz); δ_{C} 58.6, second (**1-b**) was 2-formyl-butane thiosulfinyl group: from the

methylene at δ_{H} 3.30 (2H, d, $J=7.5$ Hz); δ_{C} 47.4, next to both the methine at δ_{H} 3.07 (1H, t, $J=7.5$ Hz); δ_{C} 35.6, and the formyl function at δ_{H} 9.80 (1H, d, $J=10, 3$ Hz); δ_{C} 197.2, further to the methine at δ_{H} 2.99 (1H, t, $J=7.5$ Hz); δ_{C} 36.4, next to the methylene at δ_{H} 3.30 (2H, d, $J=7.5$ Hz); δ_{C} 44.1, and third (**1-c**) was propenyl group: from the methyl at δ_{H} 1.98 (3H, d, $J=1.75$ Hz); δ_{C} 17.5,

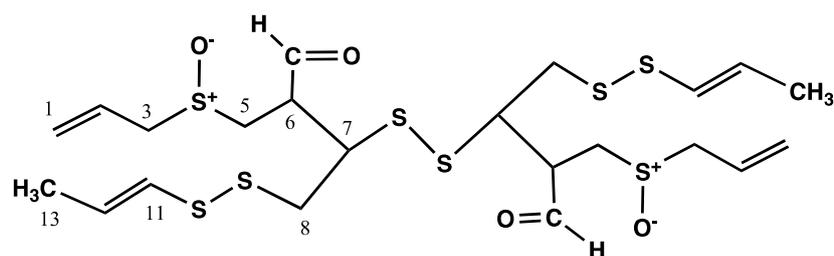


Figure 2. Structure of garlicnin L-5 (**1**).

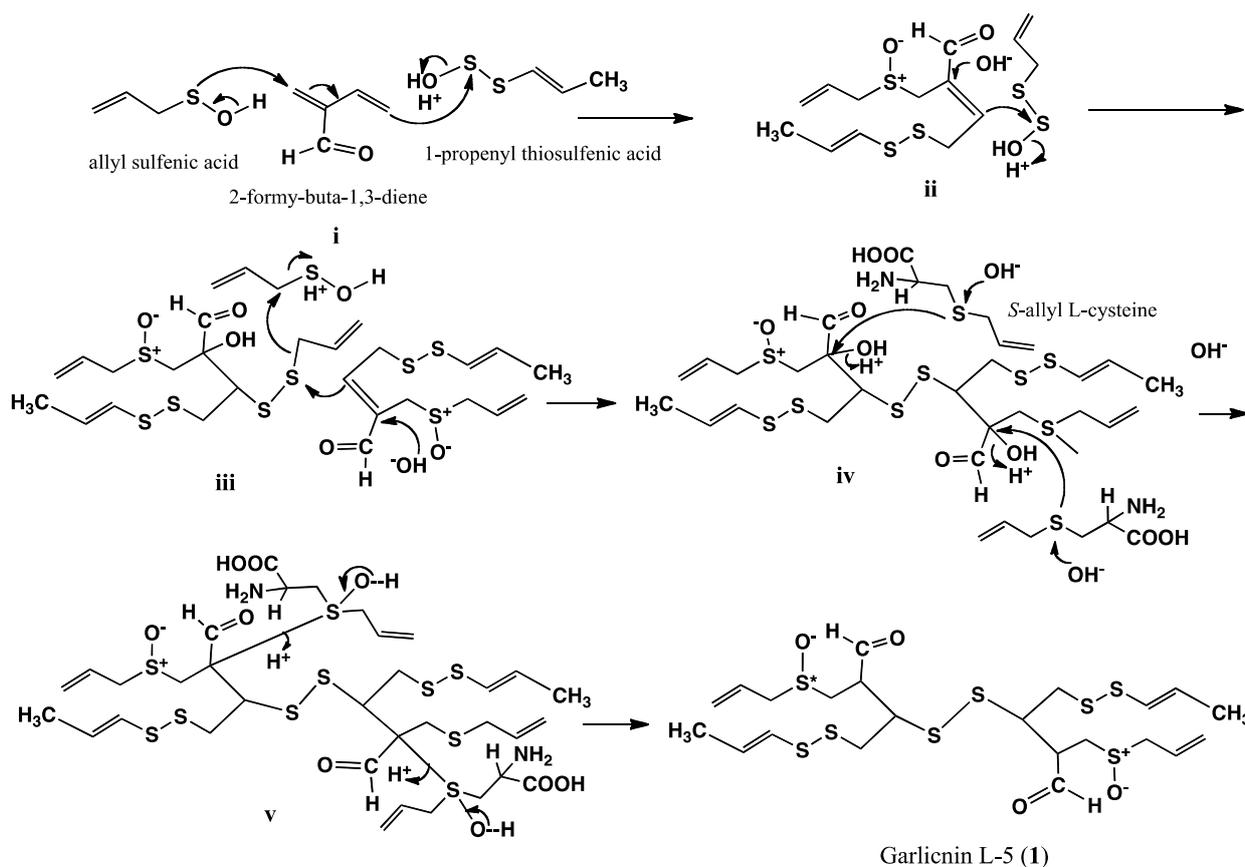


Figure 3. Hypothetical biogenesis of garlicnin L-5 (**1**).

next to the olefinic methine at δ_{H} 6.96 (1H, q, $J=6.85$ Hz); δ_{C} 145.9, further to the olefinic methine at δ_{H} 6.32 (1H, d, $J=16.6$ Hz); δ_{C} 129.2, as shown in Figure 1. Thus, the combinations between the sulfinyl group in **1-a** and the methylene group at C-5 in **1-b**, and between the thiosulfenyl group at S-9, 10 in **1-b** and the olefinic group at C-11 in **1-c**, were revealed. The remaining C-7 was considered to combine with the same molecule through the thiosulfenyl function because it appeared to form a dimer at δ_{C} 36.4; δ_{H} 2.99, t, $J=7.8$ Hz. Therefore, the structure of **1** was represented as shown in Figure 2. The generation of garlicnin L-5 (**1**) was hypothesized as shown in Figure 3, that is, i Combination of allyl sufenic acid and 1-propenyl thiosulfenic acid, which were derived from alliin, and 2-formyl-buta-1,3-diene to form **ii**, which then react with allyl thiosulfenic acid to give compound **iii**. Next, compound **iii** reacts with another sulfide (**ii**) by the withdrawal of allyl sufenic acid to form a dimer **iv**. Next, in the reaction processes in **iv** and **v**, two hydroxyl functions at C-6, C-6' would be reduced with the aid of *S*-allyl L-cysteine to afford garlicnin L-5 (**1**). 2-Formyl-butane may be derived from 2-(hydroxymethyl)butane-1,4-diol.

CONCLUSION

We isolated one acyclic sulfide, garlicnin L-5, from garlic and characterized its structure. This compound was a novel dimeric sulfide which consisted of three substructures of allyl moiety, 2-formyl-butane moiety, and propenyl moiety through the sulfinyl and thiosulfenyl bonds.

CONFLICT OF INTEREST STATEMENT

The authors declare no conflict of interest.

REFERENCES

1. El-Aasr, M., Fujiwara, Y., Takeya, M., Ono, M., Nakano, D., Okawa, M., Kinjo, J., Ikeda, T., Miyashita, H., Yoshimitsu, H. and Nohara, T. 2011, *Chem. Pharm. Bull.*, 59, 1340-1343.
2. Nohara, T., Kiyota, Y., Sakamoto, T., Manabe, H., Ono, M., Ikeda, T., Fujiwara, Y., Nakano, D. and Kinjo, J. 2012, *Chem. Pharm. Bull.*, 60, 747-751.
3. Nohara, T., Fuiiwara, Y., Ikeda, T., Murakami, K., Ono, M., Nakano, D. and Kinjo, J. 2013, *Chem. Pharm. Bull.*, 61, 695-699.
4. Nohara, T., Fuiiwara, Y., Ikeda, T., Yamaguchi, K., Manabe, H., Murakami, K., Ono, M., Nakano, D. and Kinjo, J. 2014, *Chem. Pharm. Bull.*, 62, 477-482.
5. Ono, M., Fujiwara, Y., Ikeda, T., Pan, C., El-Aasr, M., Lee, J. H., Nakano, D., Kinjo, J. and Nohara, T. 2017, *Chem. Pharm. Bull.*, 65, 102-106.
6. Nohara, T., Fujiwara, Y., El-Aasr, M., Ikeda, T., Ono, M., Nakano, D. and Kinjo, J. 2017, *Chem. Pharm. Bull.*, 65, 209-217.
7. Nohara, T., Ono, M., Nishioka, N., Masuda, F., Fujiwara, Y., Ikeda, T., Nakano, D. and Kinjo, J. 2018, *J. Nat. Med.*, 72, 326-331.
8. Nohara, T., Ono, M., Nishioka, N., Masuda, F., Fujiwara, Y., Ikeda, T., Nakano, D. and Kinjo, J. 2018, *J. Nat. Med.*, 72, 335-341.
9. Nohara, T., Ono, M., Yamaguchi, K., Sakamoto, N., Fujiwara, Y., Ikeda, T., Nakane, H., Nakano, D. and Kinjo, J. 2018, *Curr. Top. Phytochem.*, 14, 33-38.
10. Nohara, T., Fujiwara, Y., Ono, M., Ikeda, T., El-Aasr, M., Nakano, D. and Kinjo, J. 2018, *Curr. Top. Phytochem.*, 14, 87-98.
11. El-Aasr, M., Fujiwara, Y., Takeya, M., Ikeda, T., Tsukamoto, S., Ono, M., Nakano, D., Okawa, M., Kinjo, J., Yoshimitsu, H. and Nohara, T. 2010, *J. Nat. Prod.*, 73, 1306-1308.
12. Nohara, T., Fujiwara, Y., Kudo, R., Yamaguchi, K., Ikeda, T., Murakami, K., Ono, M., Kajimoto, T. and Takeya, M. 2014, *Chem. Pharm. Bull.*, 62, 1141-1145.
13. Nohara, T., Ono, M., Ikeda, T., Fujiwara, Y., Nakano, D. and Kinjo, J. 2018, *Curr. Top. Phytochem.*, 14, 71-75.
14. Fujiwara, Y., Horlad, H., Shiraishi, D., Tsuboki, J., Kudo, R., Ikeda, T., Nohara, T., Takeya, M. and Komohara, Y. 2016, *Mol. Nutrit. Food Res.*, 60, 2467-2480.
15. Tsuboki, J., Fujiwara, Y., Horlad, H., Shiraishi, D., Nohara, T., Tayama, S., Motohara, T., Saito, Y., Ikeda, T., Takaishi, K., Tashiro, H., Yonemoto, Y., Katabuchi, H., Takeya, M. and Komohara, Y. 2016, *Scientific Reports*, 6, 29588.

-
16. Block, E. and Ahmad, S. 1984, *J. Am. Chem. Soc.*, 106, 8295-8296.
 17. Fukaya, M., Nakamura, S., Nakagawa, R., Nakashima, S., Yamashita, M. and Masuda, H. 2018, *Org. Lett.*, 20, 28-31.
 18. Gao, S.-H., Nohara, T., Ikeda, T., Zhou, J.-R. and Yokomizo, K. 2020, *Curr. Top. Phytochem.*, 16, 115-121.
 19. Block, E., Dethier, B., Bechand, B., Cotelesage, J. J. H., George, G. N., Goto, K., Pickering, I. J., Rengifo, E. M., Sheridan, R., Sneed, E. Y. and Vogt, L. 2018, *J. Agric. Food Chem.*, 66, 10193-10204.
 20. Štefanová, I., Zápál, J., Moos, M., Kuzma, M. and Kubec, R. 2019, *J. Agric. Food Chem.*, 67, 9895-9906.
 21. Hagemann, T., Biswas, S. K., Lawrence, T., Sica, A. and Lewis, C. E. 2009, *Blood*, 113, 3139-3146.
 22. Gordon, S. 2003, *Nat. Rev. Immunol.*, 3, 23-35.